



## Magnesium Hydroxide Powder; 95% MgOH

### Industrial Grade Magnesium Hydroxide; Powder

Mg(OH)<sup>2</sup> : 95% ; MgO: 65.645%; Mg: 40.04%

### Typical Physical Analysis:

Specific Gravity:	2.42 g/cm <sup>3</sup>
Moisture:	<0.1
pH	9-10

### Particle Size can be produced to meet specifications

Size distribution of the granules:

Median Particle Size:	>10 microns
Passing 325 Mesh:	99%

### Typical Chemical Analysis:

<b>Magnesium Hydroxide:</b>	<b>&gt;95%</b>
CaO	<3%
SiO <sub>2</sub>	<3%
Fe <sub>2</sub> O <sub>3</sub>	0.3
Al <sub>2</sub> O <sub>3</sub>	0.1
MnO	<1
P <sub>2</sub> O <sub>5</sub>	0.1
Na <sub>2</sub> O	0.02
K <sub>2</sub> O	0.02

### Available Packaging:

50# bags & 2,000# bags  
Liquid Slurry available in some locations

### Specifications for Liquid (slurry):

The dry powder is added to water to produce a Liquid MgOH<sup>2</sup>. There is no rehydration necessary and mixing time is only required depending upon percent solids being made. A 50% solution has the viscosity of a thin latex paint. Recirculation or mechanical mixing is required for bulk storage. The MgOH<sup>2</sup> slurry can be made up to approx 55% solids. Approximately 12 dry tons per bulk liquid shipment.



### CO<sub>2</sub>Footprint associated with MgOH<sup>2</sup>:

<b>100 lb Brucite (Natural source) =</b>	<b>0.84 lbs of CO<sub>2</sub></b>
100 lb Magnesite (MgCO <sub>3</sub> ) =	63.7 lbs of CO <sub>2</sub>
100 lb Brine/Seawater (Synthetic) =	38.0 lbs of CO <sub>2</sub>

### Typical Slurry Analysis:

**50% liquid Solution**  
11.8 lbs per gallon (wet weight)  
5.95 Dry lbs MgOH<sub>2</sub> per gallon  
1.48 specific gravity